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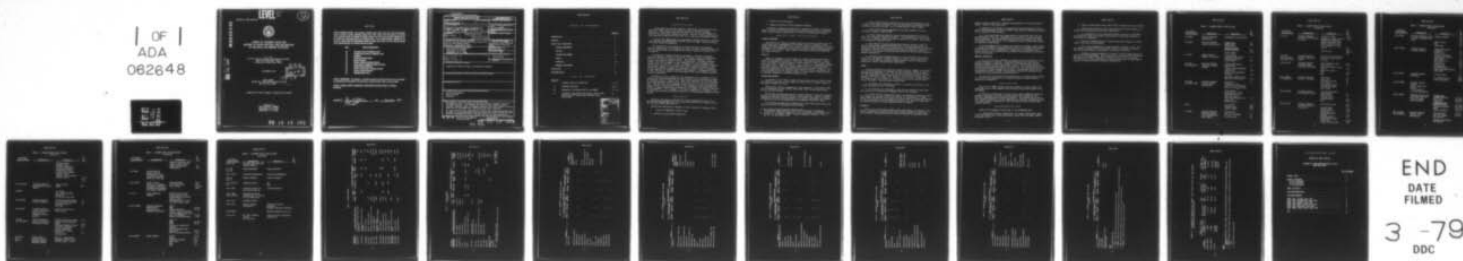
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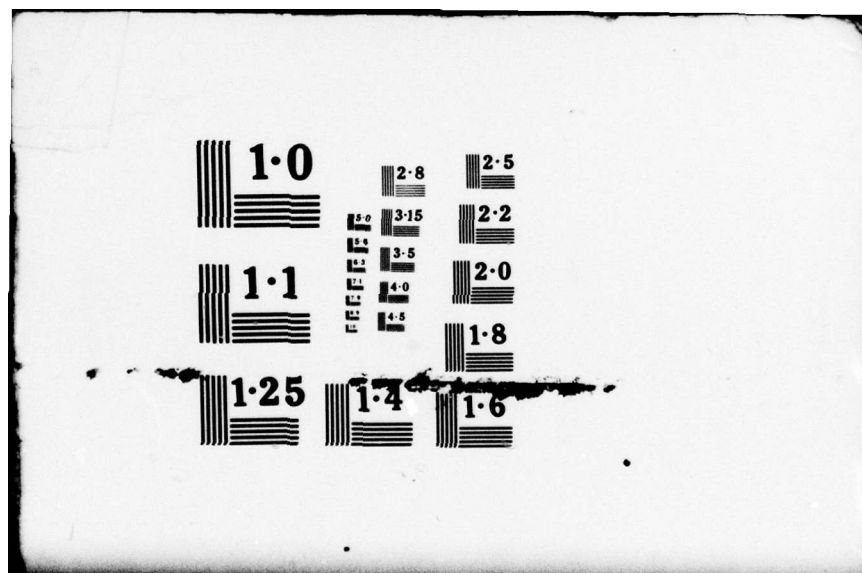
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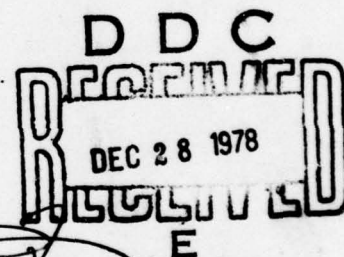


## SURVEY OF CLEANERS USAGE AND METHODS OF WASTE DISPOSAL AND RECLAMATION AT THE NAVAL AIR REWORK FACILITIES

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## INTRODUCTION

A large number of cleaning compounds representing a broad range of inorganic and organic chemicals are used by the Navy in diverse cleaning operations. Should these chemicals find their way into the environment, unfavorable and deleterious effects may result. Certain of these materials will cause immediate problems and others will cause long term environmental damage.

It is for this reason that the investigation reported herein was initiated under AIRTASK A340-0000/001B/6F57-572-401, Work Unit VQ301, Pollution Control in Aircraft Materials/Components.

The objective of this program is to abate pollution by reclamation, material substitution, or by reducing the amount of waste cleaner effluent that must be treated before final disposal into such a facility as stream, river or landfill.

In nearly all cleaning operations performed at the Naval Air Rework Facilities (NARFS) surveyed in this report, the part cleaned is flushed with water to rinse off the residual wastes. This effluent composed of dirt, cleaner, and enormous volumes of water is channelled to the waste treatment facility and after treatment is then discharged into the environment via a stream, river, or the like. However, treatment is largely ineffectual because the enormous volumes of water cause such a great degree of dilution of the cleaner that chemical treatment for breakdown, conversion and/or reclamation of the components is extremely difficult and expensive. Although the dilution is high, the adverse effect on the environment still exists through concentration by accumulation.

The impact of waste cleaners on the environment can be lessened by eliminating those cleaners performing the same function and by minimizing the number of different types of cleaners used. This latter approach can be accomplished by determining which cleaners can be substituted for by non (or less) pollutive ones. The selection of a recommended cleaner of equal or superior performance would be based on the probable impact of the components on the environment and methods of waste treatment from a standpoint of economics and chemical reactivity of the components. If a component of a cleaner formulation is hazardous, substitution by an analog with similar chemical, but non-hazardous properties should be made.

## APPROACH

Surveys of the types of cleaners and cleaner formulations used by the military services and specifically the types and quantities of cleaners used by the individual NARFS were conducted.

Studies were made of each component in these cleaners to determine their:

1. Pollutive or hazardous nature, if any.
2. Function in the cleaner formulation



3. Effect on the environment.
4. Chemical activity for waste disposal treatment.

In addition, studies were made aimed at reducing the number of different types of cleaners performing the same or equivalent function, and to determine which cleaners could be substituted for by non or less pollutive cleaners.

## RESULTS AND DISCUSSIONS

### CLEANER COMPONENTS

The study of all types of cleaners used by the military indicate that the formulations utilize approximately 14 different types of inorganic and 60 organic chemicals, among which, are 6 types of surface active agents or synthetic detergents. This represents a spectrum of 41 Federal, Military Specifications and proprietary cleaners. These cleaners with their function and components are listed in Table 1.

### TOXICITY

Among the components of the cleaners are several chemicals which must be rated as non-acceptable materials. These are: Potassium chromate, a recognized carcinogen\*\* and sodium cyanide, a deadly poison.\* Other cleaner components with toxicity hazard ratings\* of 3 are: Oxalic acid, aluminum trichloride and diethyl amine. Additional cleaner components such as sodium hydroxide, potassium hydroxide, and triethanolamine have toxicity hazard ratings of 3 and 2 in the concentrated form; however, their levels of toxicity are greatly reduced immediately prior to disposal because of their extreme solubility even if no chemical breakdown or conversion process is used.

### CLEANER REPLACEMENT

An analysis of the cleaner systems and their applications in Table 1 indicate possible reduction in their number by substitution of cleaners capable of the same or similar functions.

Specifically, based on examination of the components, cleaner MIL-C-43616 can be substituted for MIL-C-11090, MIL-C-25769, and for some applications, P-C-444 and MIL-C-22542A.

MIL-C-43616 "Cleaning Compound for Aircraft Surfaces" is used for cleaning painted and unpainted aircraft surfaces. This cleaner is a superior cleaning compound for aircraft surfaces because of its greater efficiency, cost, method of application and is environmentally more acceptable than PC-444 Type 2.

The following describes the disadvantages of each substituted cleaner and justification for its replacement by MIL-C-43616:

- \* The reference source used for toxicity hazard ratings is "Dangerous Properties of Industrial Materials: by N. Irving Sax, Fourth Edition 1975
- \*\* Listed as a carcinogen by OSHA - Chemical Regulation Reporter - Vol. 2, No. 16 - 21 July 1978 p. 629

1. MIL-C-11090 "Cleaning Compound, Solvent, Self Emulsifying" is used to remove oils, greases, asphalts, tars and preservative type materials from metallic and painted surfaces. It is a harsh and more costly cleaner.
2. MIL-C-25769 "Cleaning Compound for Aircraft Surfaces" is not as efficient a cleaner as MIL-C-43616.
3. PC-444 "Cleaning Compound, Solvent Grease Emulsifying" is used for heavy duty cleaning on unpainted surfaces. Its disadvantage is that it consists of a 2 component system that must be mixed prior to usage. MIL-C-43616 was formulated to replace P-C-444.
4. MIL-C-22542A "Cleaning Compound High Pressure Cleaner, Liquid" is used in high pressure steam cleaning machines for cleaning painted and unpainted aircraft surfaces and to neutralize acid in specialized paint stripping operations. However, steam cleaning of aircraft with high heat, pressure, and alkali is too severe a process and MIL-C-43616 can fulfill this function at ambient temperatures.
5. For Engine Cooling System Cleaners, it was determined that cleaner MIL-C-10597 can replace O-C-00432 because the former is more effective as a cleaner, and in addition, possesses conditioning and corrosion inhibiting properties. Table I contains these formulations.

#### SURVEY

A survey was made of the Naval Air Rework Facilities (NARFS) to determine what types and quantities of cleaners are used per annum, methods of waste disposal and reclamation if any. The types and quantities are given in Table II. The components in these cleaners and the toxic hazard rating of each component are listed in Table III. However, it is felt that in some instances the survey is not all inclusive but it represents the major cleaners being used at the NARFS. The survey does indicate that:

1. No efforts are being made to recover or reclaim the effluent of the cleaning compounds at any of the NARFS.
2. The effluents are discharged to some type of industrial waste treatment facility for processing or breakdown before final disposal.
3. Aircraft surface cleaner MIL-C-43616, an extremely efficient and effective cleaner can very well fulfill and surpass the requirements of cleaner MIL-C-11090, MIL-C-25769, MIL-C-22542, and also P-C-444 for certain applications. By these replacements, the environmental pollution load would be lessened since the different types of cleaner components would be reduced.

#### ECONOMICS

Based on the total usages of MIL-C-25769, MIL-C-22542, and P-C-444 by the NARFS, a comparison of the costs for these cleaners indicated a significant cost savings of approximately \$66,270 would be achieved by using MIL-C-43616



wholly in place of the first 2 cleaners and partially for P-C-444 and MIL-C-22542A. Table IV lists these data.

In addition to the savings, the substitution of MIL-C-43616 for P-C-444 Type II would eliminate the use of cresylic acids which are hazardous pollutants and the problem of their disposal.

It is to be noted that the savings quoted in this report hold true for the time period, April 1977, when these prices prevailed. Prices fluctuate rather frequently. For some cleaners there may be a decrease and for others an increase at one time period. At a subsequent period an appreciable change in trend may be experienced.

If the recommendation for cleaner substitution is followed, large volumes of cleaner MIL-C-43616, including the sum total of the volumes of the substituted cleaners, would be purchased as a single item. By purchasing high volumes the cost of the cleaner would be reduced, thus ensuring a cost saving just on volume alone.

#### CHROMATE INHIBITORS

It is to be noted that inclusion of chromate corrosion inhibitors to the cleaner formulation is optional. Hexavalent chromium, the type found in potassium chromate, constitutes a hazard to the environment and presents a difficult problem in waste treatment and disposal. The conventional methods of extracting chromium from water solution are, the ion exchange process, where the solution is passed through a bed or beds of ion exchange resins and the precipitation process, in which the hexavalent chromium is converted to the trivalent form and subsequently precipitated out.

In view of the low chromium concentration due to the high dilution resulting from the water flushing in the cleaning operation, neither method is practical nor economical.

#### CONCLUSIONS

Each of the 5 NARFS surveyed indicated treatment of spent cleaner was being made prior to final disposal thus reducing the pollutive impact on the environment.

No efforts are being made to recover waste cleaner components since the methods of application of these cleaners preclude any economic and anti-pollutive advantages gained through reclamation. The substitution of non or less pollutive cleaners or cleaner components for pollutive and hazardous ones is a viable option to minimize or reduce pollutive and also to effect some economic benefits.

#### RECOMMENDATIONS

Based on the findings of this study, it is recommended that:

1. Cleaner MIL-C-43616 be substituted for cleaners MIL-C-11090, MIL-C-25769, MIL-C-22542 and in some applications, other than some NARF's special preference P-C-444.

2. Engine cooling system cleaner MIL-C-10597 be substituted for O-C-00432.
3. Chromates be replaced by other less polluting corrosion inhibitors if inhibitors are required in the cleaner formulation.
4. In consideration of the cleaning procedures practiced at the NARFS which generally result in extremely high dilution of the waste cleaners, no reclamation is recommended.
5. All waste cleaners should be treated before final disposal to reduce adverse effects on the environment.
6. Due to the incompleteness of some usage data obtained, an all inclusive survey of cleaners used at the NARFS should be conducted to determine what other substitution or replacements of cleaners can be made.
7. Records on usage and methods of disposal of waste cleaners should be kept by the individual NARFS. They would serve as an indication of the nature and extent of the impact these waste cleaners may have on the environment. They would also aid in determining the possible cancellation of some cleaners in favor of others having less pollutive formulations.

TABLE I. CLEANERS USED BY THE MILITARY

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
O-C-00430	Cleaning compound paint brush	Information not available	
O-C-00432	Cleaning compound engine cooling system	<u>Cleaner part</u> Oxalic acid Wetting agent <u>Conditioner part</u> Sodium carbonate Wetting Agent	98 min.  98 min.
P-C-435A	General purpose (powder, heavy duty)	Alkyl benzene sulfonate (40% active) Sodium metasilicate $\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$	10 90
P-C-437A (MS 36423 (MU))	Cleaning compound high press (steam) cleaner	Sodium metasilicate $\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$ Primary sodium phosphate $\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$ Sodium tripolyphosphate $\text{Na}_5\text{P}_3\text{O}_{10}$ Non-ionic surface active agent mild 16791 Type I	35 10.5 52.5 2.0
P-C-444A (MS 36424 (MU))	Cleaning compound, solvent, grease emulsifying	Kerosene C Dispersant N-10 (non-ionic) condensation product of $\text{CH}_2 - \underset{\text{O}}{\text{C}} - \text{CH}_2$ and alkyl phenol potassium resin soap potassium tall oil fatty acid soap free acid calculated as oleic acid hexylene glycol	52  1.5
Type 2		Pet solvent cresylic acid caustic potash	62% 20% < 5%
P-C-446A	Cleaning Compound solvent detergent (for cleaning coils and filters)	$\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$ Non-ionic detergent De-ionized $\text{H}_2\text{O}$	5 10 85



TABLE I. CLEANERS USED BY THE MILITARY  
(continued)

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
P-C-535	Cleaning compound, platers electrocleaning for steel	Silicate as $\text{SiO}_2$ Phosphate as $\text{P}_2\text{O}_5$ (complex phosphates) anhydrous organic detergent (100% active basis) as alkyl aryl sulfonate or alkyl sulfate  Caustic soda	10-35 5 min.  0.4 min. Remain- der
O-C-1824 (MIL-C-20207C) (MS 36435 (MU))	Cleaning compound, solvent, heavy duty, liquid	Must not have chlorinated solvent or other toxic solvents	
MIL-C-5410B	Cleaning compound, aluminum surface non-flame-sustaining	Ortho phosphoric acid Citric acid Non-ionic surface active agent MIL-D-16791 Type I MEK $\text{H}_2\text{O}$	23.9 11.9 10.2 15.3 38.7
MIL-C-6864C (MS 36426 (MU))	Cleaning compound, solvent, oil-cooler	95% ethanol Cresol Usp Methylene Chloride Pot. oleate Polyethylene glycol monalkylaryl ether MIL-D-16791	10 10 70 8 2
MIL-C-25769H	Cleaning compound for aircraft surfaces	Trisodium phosphate Dodecahydrate Triton-X-100 Ethylene glycol monoethylether $\text{H}_2\text{O}$	10 2.0 6.0 82.0
MIL-C-27251A	Cleaning compound aircraft surface low temperature	Petroleum solvent  Diethyl amine Triethanolamine Oleic acid Sodium heptadecyl- sulfate Ethylene glycol Fatty alkylol amide condensate t-alkyl primary amine	67.95 $\pm 5$ .73 1.48 5.63 1.56 .90 18.26 3.48

TABLE I. CLEANERS USED BY THE MILITARY  
(continued)

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
MIL-C-38736	Cleaning compound solvent for use in integral fuel tanks	Aromatic naphtha  Ethyl acetate MEK Isopropanol	50% by vol. 20 20 10
MIL-C-43616B	Cleaning compound aircraft surfaces	Tetrapyropotassium phosphate $\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$ Potassium chromate Ethylene glycol mono- butyl ether KOH (45% by wt.) Oleic acid Solvesso 150 Cyclohexanol Monoethanolamine Dodecyl benzene sulfonic acid (85%)	2.5 0.7 0.1 5.0 4.2 8.3 41.7 2.8 6.7 7.2
MIL-C-51340	Cleaning compound electrical	Silicone fluid  MIL-S-81087 Type II Trichloroethylene	By vol. 50+ .5 50+ .5
MIL-C-81302B	Cleaning compound solvent Trichlorotrifluoro- ethane 1,1,2-Trichloro 1,2,2-Trifluoroethane		
MIL-C-10597D	Cleaning compound with conditioner and inhi- bitor for engine cooling systems	<u>Cleaner part</u> Aluminum chloride Oxalic acid Cornstarch <u>Conditioner part</u> Sodium silicate Trisodium phosphate Borax Sodium carbonate	14.6-15.2 84.3-85.5 0.15-0.3 51.9-63.5 22.1-23.8 18.3-19.8 7.9-9.5
MIL-C-11090 (MS 36436 (MU))	Cleaning compound solvent, self- emulsifying	10% high flash coal tar naphtha 30% kerosene  Detergent concentrate Cyclohexanol	100 pts. by vol.



TABLE I. CLEANERS USED BY THE MILITARY  
(continued)

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
		Ethylene glycol monobutyl ether Diacetone-diethylene triamine condensate Ethanolamine oleate in high flash naphtha Diglycol oleate aromatic amine sulfonate Sorbitan oleate	12.25 by vol.
MIL-C-22542A	Cleaning compound, high pressure cleaner, liquid	Caustic potash $K_2CrO_4$	2 <.5
OOPH31D		(1) Isopar boil range 405-485°C Humble oil (2) Unitol fatty acid	
MIL-C-81964	Cleaning compound avionic components	Trichlorotrifluoroethane Silicone fluid VV-D-1078 Propellant Type 12 BB-F-1421	74.7 .3 25
MIL-C-372B	Cleaning compound, solvent (for bore of small arms and automatic aircraft weapons)	Kerosene, butyl cello- solve, water	
P-C-436 (MS 36422(MU))	Cleaning compound, alkali, ferrous and non-ferrous surface	Sodium metasilicate anhyd. Primary sodium phosphate anhyd. Trisodium phosphate anhyd. Non-ionic surfactant Avionic surfactant sodium alkyl benzene sulfonate $C_{12}$ - $C_{18}$ alkyl	31.3 12.3 24.5 1.9 23.7
TT-T-291E Type I Type II	Thinner paint volatile spirits, petroleum spirits Bearings cleaner	Type II - constituents allowable compounds with olefinic unsat. - trace	

TABLE I. CLEANERS USED BY THE MILITARY  
(continued)

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
		Aromatic compounds with eight or more C atoms except ethylbenzene	8%
		Ethylbenzene	20%
VV-L-800A	Lubricating oil general purpose preservative (water displacing low temp) flash pt. 275°F min.		
MIL-S-18718	Safety solvent general use cleaning and grease removing of assembled and unassem- bled engine components	Dichloromethane Perchloroethylene Min. spirits	25+1 5+.5 to make 100%
P-C-111	Carbon removing compound	Ethylene glycol mono- butyl ether (3 gals.) Diethyleneglycol diethyl ether (1 gal.) Monoethanolamine (3 pts. to every 45 gals. H <sub>2</sub> O)	
MIL-C-14460B	Corrosion removing NaOH base for electrolytic or immersion application	Type I NaOH Sodium gluconate Trisod. salt of N-hydroxy ethylene diamine-triacetic acid (3 NaEDTH-OH) Others including foamers	54 max. 25 min.  13 min. 7 max.
		Type II NaOH NaCN Chelate or sequestrant compound EDTA tetrasodium salt dehydrate) Others including foamers	35 min. 25-35  25-35 4 max.
MIL-C-19853	Carbon remover	Cresol Phenol Methylene chloride Oleic acid KOH EDTA	33.62 1.0 54.9 6.8 3.5 .16

TABLE I. CLEANERS USED BY THE MILITARY  
(continued)

<u>Government Specification</u>	<u>Designation</u>	<u>Components</u>	<u>% By Wt.</u>
MIL-R-7751	Silicate type paint and varnish remover		
O-S-598 Type II	Sodium hydroxide	Sodium hydroxide	
MIL-P-11970	Potassium Permanganate	Potassium Permanganate	
O-S-571 Type II	Sodium carbonate	Sodium carbonate	
Turco SMUTGO NC	Aluminum cleaner	HF HNO <sub>3</sub>	
Turco 3823	Aluminum cleaner hot tank carbon remover	O-Dichlorobenzene	
Turco 4008 (MIL-D-26549)	Alkaline descaling compound for hot section of jet engine parts		
Turco 4213	Aluminum cleaner		
Turco 5873	Radome stripper non-phenolic	Methylene chloride thickener Chromate corrosion inhibitor	
MIL-D-23962		Chevron absorption oil No. 2	
Turco 4215	Hot tank aluminum cleaner Non-silicated	Sodium nitrate, detergents, emulsifiers	



TABLE II. CLEANING COMPOUNDS

Government Specification	Cleaning Compound (Description)	Consumption per Annum (Gallons)					Total Used by NAFS
		Alameda	Cherry Point	Jacksonville	Norfolk	North Island	
MIL-C-372B	Solvent for bore of small arms automatic aircraft				500		500
MIL-C-5410B	Aluminum surface cleaner non-flame sustaining		300		400	660	1,940
MIL-C-6864C	Solvent oil cooler					660	740
MIL-R-7751	Remover - paint and varnish	54 <sup>2</sup>		440			494
MIL-D-9063	Desalant integral fuel and oil tanks	3,545 <sup>2</sup>			6,600		10,145
MIL-P-11970	Potassium permanganate	3,040 lbs <sup>1</sup>					3,040 lbs <sup>1</sup>
MIL-C-14460	Corrosion removing NaOH base - electrolyte immersion	2,500 <sup>2</sup>		2,500 <sup>2</sup>	5,280		10,280
MIL-D-19853A	Carbon removing	9,185			8,360	660	18,105
MIL-D-22542A	High press cleaning liquid	1,364 <sup>2</sup>	660	8,800		900	12,824
MIL-P-23962	Chevron absorption oil No. 2	2,257					2,257
MIL-C-25769	Aircraft surface		2,660	870 <sup>2</sup>	3,300		6,930
MIL-D-26549 (Turco 4008)	Alkaline descaling of hot engine parts	2,805					2,805
MIL-C-43616	Aircraft surface					36,000	44,040
MIL-C-81294	Epoxy paint remover	53,481			40	17,600	71,081

TABLE II. CLEANING COMPOUNDS  
(continued)

Government Specification	Cleaning Compound (Description)	Consumption per Annum (Gallons)					Total Used by NARS
		Alameda	Cherry Point	Jacksonville	Norfolk Island	Pensacola	
MIL-R-81835	Remover organic coating hot tank type						4,440
OS-571 Type II	Sod, carbonate	1,430 lbs <sup>1</sup>					1,430 lbs <sup>1</sup>
OS-598 Type II	Sod, hydroxide	5,200 lbs <sup>1</sup>					5,200 lbs <sup>1</sup>
PC-111	Carbon removing	14,520				220	14,740
PC-444A	Solvent grease emulsifying		220	3,300	5,060	5,060	45,640
PC-535	Platers electrocleaning for steel		220	112 lbs <sup>1</sup>		1,800	2,020 plus 112 lbs
Turco 3823	Aluminum cleaner hot tank carbon remover	5,170					5,170
Turco 4213	Aluminum cleaner				800		
Turco 4215	Hot tank aluminum cleaner non-silicated	2,250 lbs <sup>1</sup>			1,200 lbs <sup>1</sup>		3,450 lbs <sup>1</sup>
Turco 5873	Radome stripper	2,255					2,255
Turco Smutgo	Aluminum cleaner				660 lbs <sup>1</sup>		660 lbs <sup>1</sup>
Calgon-SW3						3,000	3,000
Neoprene Stripper							
GACO SR-14					600		600
S-14							
B&B 50-50							
NOTE: (1) Data submitted in pounds (2) Data submitted in pounds and converted to gallons							
TOTAL							260,000 Gals. plus 13,892 lbs



TABLE III. COMPONENTS OF CLEANERS USED BY THE NAFES

Component	Toxic Hazard Rating						Comments
	Chronic Systemic	Inhalation	Ingestion	Skin Absorption	Acute Systemic	Skin Absorption	
Alkyl sulfate							Acute inflammatory and corrosive to tissue.
Alkyl benzene sulfonate							No data
Aluminum chloride	U	U	U	U	U	U	No data
Amine sulfonate							More toxic than glycols
Butyl cellosolve							May be carcinogenic
100 SUS coastal oil							Innocuous
Cornstarch							Innocuous
Citric acid							
Cresol	2	2	2	2	2	2	
Cresylic acids	2	2	2	2	2	2	
Cyclohexanol							
Cyclohexanone	1		1		1	1	A suspected carcinogenic
Diacetone alcohol	U	U		U	2		
O-dichlorobenzene		2			2		
Diethylenetriamine	U	U	2	U	2	2	

TABLE III. COMPONENTS OF CLEANERS USED BY THE MARPS  
(Continued)

Component	Toxic Hazard Rating					Comments
	Chronic Systemic		Skin		Acute Systemic	
	Ingestion	Inhalation	Absorption	Inhalation	Ingestion	Skin Absorption
Diethylamine						Chronic and acute systemic toxicity unknown
Diethylene glycol diethyl ether						Toxicity unknown
Diglycol oleate						No data
Dipropylene glycol methyl ether						Low ingestion toxicity
Dispersant N-10						No data
Dedecyl benzene sulfonic acid						No data
Ethanolamine oleate						No data
Ethanolamine	U	U	U	2	2	
Ethanol	1	1	1			
Ethyl acetate	1	1	1			
Ethylbenzene	U	U	U	2	2	2
Ethyl cellosolve	1	1				
Ethylene glycol	2			3		
Ethylene glycol monobutyl ether						See butyl cellosolve
Ethylene glycol monoethyl ether	1	1				See ethyl cellosolve

TABLE III. COMPONENTS OF CLEANERS USED BY THE NAFPS  
(Continued)

Component	Toxic Hazard Rating						Comments
	Chronic Systemic			Acute Systemic			
	Ingestion	Inhalation	Skin Absorption	Ingestion	Inhalation	Skin Absorption	
Ethylene diamine tetracetic acid	1						
Glycerine				1			
Hexylene glycol	U	U	U	1	1		
Hydrofluoric acid	3	3		3	3		Very corrosive to skin
Isopropanolamine							Toxicity unknown
Kerosene	3						
Methyl ethyl ketone	U	U	U		2		
Methanol	2	2	2				
Methylene chloride (dichloromethane)	1	1	1				
Monomethyl aryl cellosolve							More toxic than glycols
Monobutyl cellosolve	2						
Monoethanolamine							See ethanolamine
Mineral spirits	U	U	U	2	2		
Naphtha (aromatic)	3	3	3				
Nitric acid	U	U	U		3		
Octyl phenoxy polyethoxy ethanol							No data



TABLE III. COMPONENTS OF CLEANERS USED BY THE NARFS  
(Continued)

Component	Toxic Hazard Rating						Comments
	Chronic Systemic			Acute Systemic			
	Ingestion	Inhalation	Skin Absorption	Ingestion	Inhalation	Skin Absorption	
Oleic acid	1						
Oxalic acid	2	2					
Perchloroethylene	2	2	2				
Phenol	2	2	2				
Phosphoric acid	U	U	U	2	2		
Paraffin oil (mineral oil)							Recognized carcinogen of skin and scrotum
Potassium chromate							Recognized carcinogen
Potassium hydroxide	U	U	U	3			
Potassium oleate							No data
Potassium resin soap							No data
Potassium tall oil fatty acid soap							No data
Polyethylene glycol monoalkyl aryl ether							No data
Pentasodium triphosphate				2			
Silicone fluid							Low toxicity

TABLE III. COMPONENTS OF CLEANERS USED BY THE NARPS  
(continued)

Component	Toxic Hazard Rating						Comments
	Chronic Systemic			Acute Systemic			
	Ingestion	Inhalation	Skin Absorption	Ingestion	Inhalation	Skin Absorption	
Sodium carbonate	U	U	U	2	2		Recognized carcinogen
Sodium chromate							Very poisonous
Sodium cyanide							Toxicity unknown
Sodium gluconate							No data
Sodium heptadecyl sulfate							Corrosive to all body tissue
Sodium hydroxide	U	U	U	U	U	U	Mild irritant
Sodium phosphate (monobasic)							No data
Sodium petroleum sulfonate							May be carcinogenic
Solvesso 150							Suspected carcinogen
Sorbitan oleate							
Tetrapotassium pyrophosphate	1	1		2	2		
Tetrasodium EDTA							
Trichloroethylene	3						Highly toxic



TABLE III. COMPONENTS OF CLEANERS USED BY THE NARFS  
(continued)

Component	Toxic Hazard Rating						Comments
	Chronic Systemic			Acute Systemic			
	<u>Ingestion</u>	<u>Inhalation</u>	<u>Skin Absorption</u>	<u>Ingestion</u>	<u>Inhalation</u>	<u>Skin Absorption</u>	
Triethanolamine	3		1				No data
Trisodium EDTA							No data
Triton X-100 (see octyl phenoxy polyethoxy ethanol)							

NOTE:

## TOXIC HAZARD RATING CODE

1. SLIGHT - causes readily reversible changes which disappear after end of exposure
  2. MODERATE - May involve both irreversible and reversible changes not severe enough to cause death or permanent injury
  3. HIGH - May cause death or permanent injury after very short exposure to small quantities
- U UNKNOWN - No information on humans considered valid by authors

TABLE IV. ECONOMICS OF REPLACING MIL-C-22542, MIL-C-25769, AND PC-444 (FOR CERTAIN APPLICATIONS) WITH MIL-C-43616<sup>2</sup>

Cleaner	Total Vol. Used at NARFS per Annum (Gals.)	Cost to Govt. <sup>3</sup> per 55 Gals.	Cost per Gal.	Cost of MIL-C-43616 per Gal.	Savings per Gal.	Savings per Total Vol. Used per Annum
MIL-C-25769	7,020	\$ 72.76	\$1.32	\$ .80	\$ .52	\$ 3,650
MIL-C-22542	12,820	\$184.04	\$3.35	\$ .80	\$2.55	\$16,350 <sup>1</sup>
PC-444	45,640 (22,820)	\$150.80	\$2.74	\$ .80	\$1.94	\$44,270 <sup>1</sup> \$66,271

## NOTE:

- 1 - Assume replacing  $\frac{1}{2}$  total volume of PC-444 and MIL-C-22542 for application that MIL-C-43616 can fulfill.
- 2 - Total volume of MIL-C-43616 used per annum at the NARFs is 44,040 gallons at \$43.87 per 55 gallon lots.
- 3 - Prices are as of April 1977.

# D I S T R I B U T I O N   L I S T

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